

PORTABLE UV PLASMA AIR PURIFIER IN AUTOMOBILE

F. L. Kow,^a H. Nasution,^{a,*}

^aAutomotive Development Centre,
Faculty of Mechanical Engineering,
Universiti Teknologi Malaysia, Johor, Malaysia.

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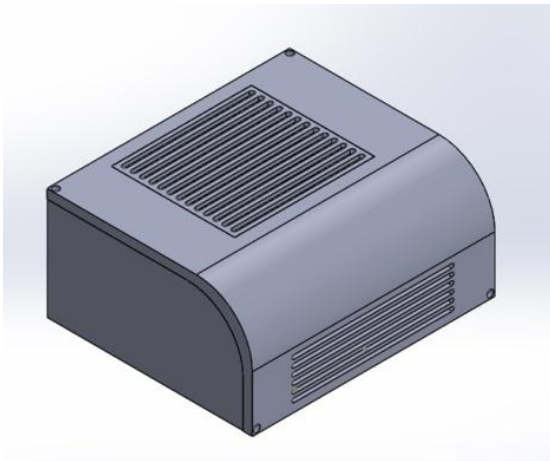
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*Corresponding author

henry@utm.my

GRAPHICAL ABSTRACT



ABSTRACT

Air quality in automobiles is worse than the air outside. As such, there is a need to develop efficient air purifier to remove air pollutants in automobiles. In this study, a prototype has been assembled and tested for its functionality and performance using air quality measuring devices. Performance test has been done to test the efficiency of the air purifier in removing five different air pollutants, which are CO₂, NO_x, CH₂, CH₃ and SO₂. The percentage change of removal of each air contaminant is quite high, having removed SO₂ with 93%, CO₂ with 48.85%, NO_x with 35.96%, CH₂ with 27.05% and CH₃ with 25.35%. Overall, the air purifier is highly efficient in removing air pollutants.

KEYWORDS

Air purifier; automobile; UV plasma; air pollutants

INTRODUCTION

The increase of industrial activities and automobiles nowadays has increased the pollution in air. Although most people tend to think that the air inside their automobiles is cleaner than the air outside, studies has found that the air quality inside the automobile is filthier compared to outdoor, as pollutants from the outside flow into the automobiles and influence the air quality in cars negatively [1,4,6].

There are many pollutants present in automobiles, such as volatile organic compounds (VOCs) [2] carbon dioxide (CO₂), sulphur dioxide (SO₂), particulate matter (PM) [7], carbon monoxide (CO), nitrogen oxides (NO_x) [8], biocontaminants and many more. Removal of these hazardous air pollutants effectively and quickly is necessary.

Thus, it is important that air purifiers be installed inside automobiles to improve the air quality. There are many types of air purifiers available in the market. There are UV (ultra-violet) air purifiers [3] and plasma technology air purifiers [5].

In this study, a UV plasma air purifier, which combines the UV and plasma technologies, is proposed for assembly and test. Other types of filters are also proposed for assembly in the air purifier to increase the efficiency of air pollutants removal of different types and sizes.

METHODOLOGY

Design, Assembly and Fabrication

The proposed air purifier was designed according to aspects of health, comfort, environment and energy consumption. After a detailed design drawing was obtained, the prototype was fabricated according to the configuration specified in the detailed design drawing. The casing of the air purifier and its holders inside were printed using a 3-D printer. After all the components were gathered, the air purifier was assembled. Figure 1 shows the detailed drawing of the final design of the prototype.

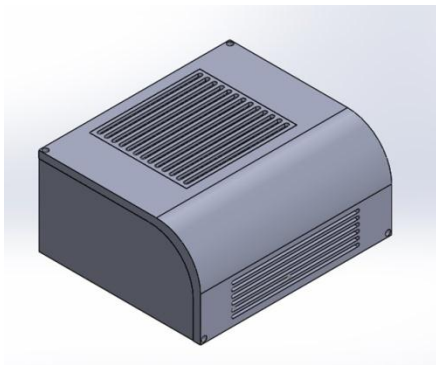


Figure 1: Detailed drawing of final design

Functional Testing

Once the prototype was completed, it was put into testing for functionality and electrical power consumption. A non-regulated DC power supply was used to power up the prototype and check for its functionality. A digital multimeter was used for continuity check on the electronic circuit and measuring the DC current consumed by the air purifier. An anemometer was then used to measure the air flow and air velocity of the air purifier.

Performance Testing

Four tests were carried out to determine the performance of the air purifier. First, a MKS Cirrus 2 atmospheric pressure gas monitor or a residual gas analyser was utilized to measure the performance of the air purifier in removing four air pollutants i.e. CO₂, NO_x, CH₂ and CH₃. The test was carried out in a wooden box of dimension 370mm x 470mm x 475mm under controlled conditions. The second test was measuring the removal of SO₂ using a YESAIR indoor air quality monitor. This test was carried out inside a closed car with extreme conditions of smoke from a burning cigarette. The

third testing condition was the testing of the effectiveness of removing CO₂ by UV plasma technology, while the fourth test was to test the effectiveness of removal of CO₂ under different airflow velocities (1.90 m/s, 2.40 m/s, 2.90 m/s). These last two tests were carried out using the box as mentioned above.

For all the tests, the concentration of the air pollutants were monitored and measured. Then, all data were plotted in graphs and the rate of change (gradient of best fit line) was measured. The percentage change was also calculated using the following equation:

$$\text{Percentage change} = (x_j - x_i) / x_i * 100 \quad (1)$$

where x_i is the initial value and x_j is the final value.

RESULTS AND DISCUSSION

Effectiveness Testing in Pollutants Removal

The testing was carried out under a temperature of 28°C and relative humidity of 74.2%. Figures 2, 3, 4, 5 and 6 show the graphs of the air pollutants (CO₂, NO_x, CH₂ and CH₃) pressure change over time.

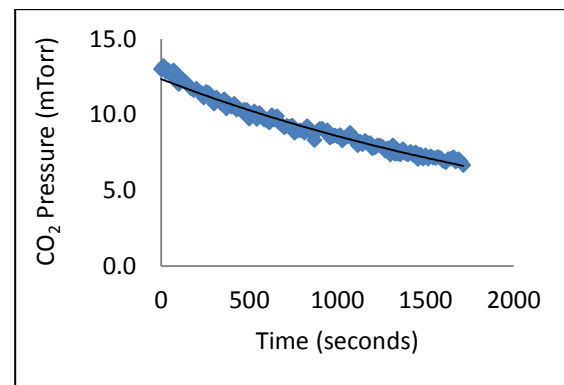


Figure 2: CO₂ pressure over time

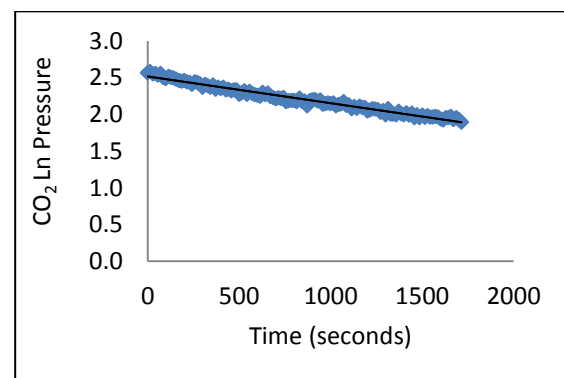


Figure 3: CO₂ ln (pressure) over time

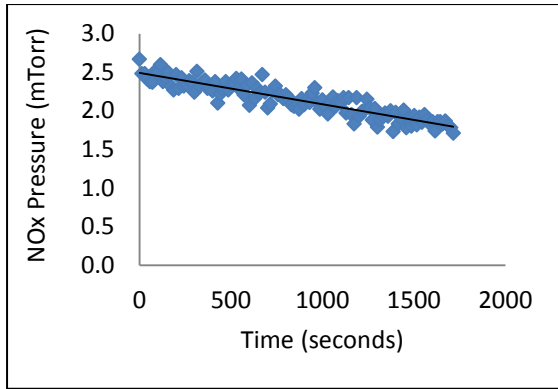


Figure 4: NOx pressure over time

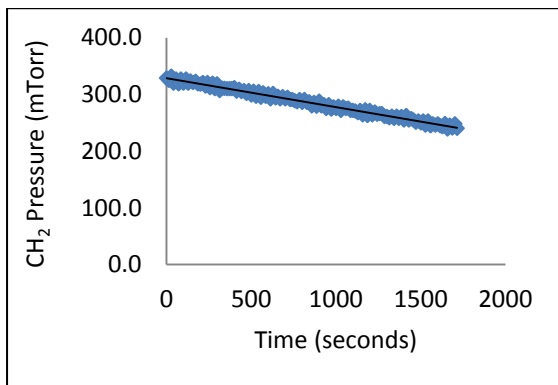


Figure 5: CH₂ pressure over time

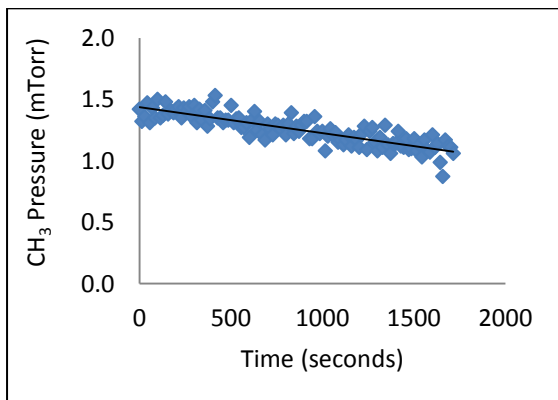


Figure 6: CH₃ pressure over time

Table 1 shows compilation of the rate of decrease and percentage change of each air pollutant. The rate of decrease of the pressure of CH₂ was the highest, while the decrease of CH₃ was the lowest. Though the rate of decrease of CH₂ was higher than the rate of decrease of CO₂, the percentage change of CO₂ was almost two times the percentage of CH₂. This testing shows that all components could be cleaned off by the air purifier. CO₂ was the most efficiently removed component by the air purifier. Overall, the air purifier had been found highly efficient in removing CO₂ compared to NOx, CH₂ and CH₃ components.

Table 1: Rate of decrease and Percentage change of CO₂, NOx, CH₂ and CH₃

Types	CO ₂	NOx	CH ₂	CH ₃
Rate of decrease (mTorr/second)	0.0004	0.0004	0.0511	0.0002
Percentage change (%)	48.85	35.96	27.05	25.35

Effectiveness of Removing SO₂

Figure 7 shows the graph of SO₂ content over time. It shows a decreasing trend of the SO₂ content after the installation of the prototype. The test was carried out under temperature of 28.8°C and relative humidity of 80.0%. The rate of decrease of the SO₂ was 0.003ppm/second while the percentage change was 92.86 %. This shows a really high efficiency in removing the SO₂ from the atmosphere, as almost 93% of the SO₂ had been removed in 15 minutes (900 seconds) with a rate of 0.003 ppm per second.

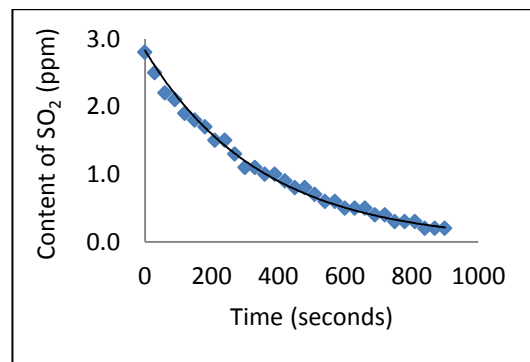


Figure 7: SO₂ content over time

Effectiveness CO₂ Content for Different Airflow Velocities

The readings of the change in CO₂ under airflow velocities 1.90 m/s, 2.40 m/s and 2.90 m/s were taken under temperature of 28.8°C and relative humidity of 80.0%. Figure 8 displays the content of CO₂ over time for different airflow velocities.

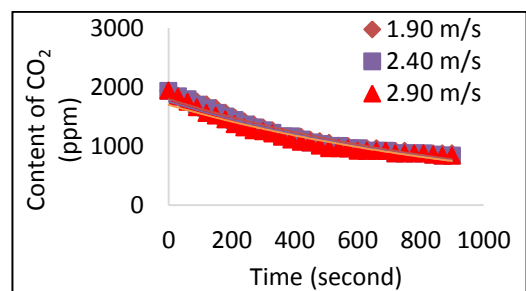


Figure 8: CO₂ content for different airflow velocities

As shown in Table 2, it was found that with higher airflow velocity, the rate of decrease in CO₂

became larger. These two parameters were proportional to each other. This shows that with higher the airflow velocity, the removal of CO₂ over time became more efficient. The percentage change also showed agreement by showing that there was an increase in removal of CO₂, with the increase of percentage changing from 2.40 m/s to 2.90 m/s, less than 1%. Based on these findings, we can conclude that the airflow velocity should be higher to ensure the efficiency of the air purifier to remove CO₂ from the intended surrounding.

Table 2: Rate of decrease and percentage change of CO₂ under different airflow velocities

Airflow velocity	1.90 m/s	2.40 m/s	2.90 m/s
Rate of decrease (ppm/second)	1.33	1.51	1.54
Percentage change (%)	54.64	56.79	56.83

CONCLUSIONS

The results prove that the air purifier can remove air pollutants found in automobiles, which are CO₂, NO_x, CH₂, CH₃ and SO₂. The air pollutants removal efficiency by the air purifier from high to low is as follows: SO₂ > CO₂ > NO_x > CH₂ > CH₃. It is also proven that the UV plasma technology combination is more efficient than utilizing only one technology. Higher airflow velocity will increase the efficiency of the air purifier in removing air pollutants from the environment.

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